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Note: Not all of the reactions will occur. For those that do not, write no reaction. 1. A piece of copper is dropped into a container of water. No reaction 2. Liquid bromine is added to a container of sodium iodide crystals.  $\text{Br}_2(l) + 2\text{NaI}(s) \rightarrow 2\text{NaBr}(s) + \text{I}_2(s)$  3. An aluminum strip is immersed in a solution of silver nitrate.

### Ultimate equation answers - Max Study

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### The Ultimate Chemical Equations Handbook by George R....

The Ultimate Chemical Equations Handbook SnCO tin(II) carbonate sodium hydrogen carbonate NaHCO 1. 2. 3. 4. 5. 6. 8. 9. 10. 11. vanadium(V) oxide H<sub>2</sub>O dihydrogen monoxide (NH<sub>4</sub>)<sub>2</sub>CO ammonium oxalate 4 224 polonium(VI) thiocyanate Po(SCN)<sub>6</sub> tetraphosphorus decaoxide P<sub>4</sub>O<sub>10</sub> 12. 13. 15. 16. 17. 18. 19. manganese(VII) oxide copper(II) dihydrogen phosphate

### East Boston High School

This ultimate chemical equations handbook answers chapter 10, as one of the most operational sellers here will utterly be among the best options to review. Ultimate Chemical Equations Handbook Answers Answers for the questions in Chapter 8: ROUND 3 Exercise 8-1: Using the activity series, predict and balance the following single replacement reactions.

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