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## Math Skills Transparency Worksheet Answers Chapter 14

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compound. 2. How many potassium ions (group 1) would be related to balance the charge of each of the following in a compound?

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The exponents are added.  $(2.00 \times 10^4) = 2.00 \times 10^{3+4} = 8.00 \times 10^7$  Division When numbers in scientific notation are divided, only the number is divided. The exponents are subtracted.  $9.60 \times 10^7 \div 1.60 \times 10^4 = 6.00 \times 10^3$  Math Handbook Transparency Masters  $9.60 \times 10^7 \div 1.60 = 6.00 \times 10^3$  Chemistry: Matter and Change Name Date Class 58 MAT" HANDBOOK  
TRANSPARENCY WORKSHEET Operations with Scientific

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Notation I. Perform the following operations and express the answers in scientific notation.

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MATH SKILLS TRANSPARENCY WORKSHEET Use with Chapter 1, Section 11.1 MATH SKILLS TRANSPARENCY WORKSHEET 16 Section 11.2 Determine the mole ratios for each of the following balanced chemical equations.  $1.2\text{C(s)} + \text{O}_2\text{(g)} \rightarrow 2\text{CO(g)}$  1 mol  $\text{O}_2$  2 mol C 2 mol CO/2 mol C 2 mol C/1 mol  $\text{O}_2$  2 mol CO/1 mol  $\text{O}_2$  2 mol C/2 mol CO 1 mol  $\text{O}_2$  / 2 mol CO 2.  $\text{WO}_3\text{(s)} + 3\text{H}_2\text{(g)} \rightarrow \text{W(s)} + 3\text{H}_2\text{O(l)}$

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