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Subject: Engineering Chemistry Subject Code: 17CHE12/22
Duration: 3 Hours Max. marks: 100 Note: Answer any FIVE full questions, choosing one full question from each module Module -I
1 a. What is single electrode potential? Derive the Nernst equation for single electrode potential . (7 marks) b. What are concentration cells? The cell potential of Cu concentration cell Cu/CuSO₄

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Engineering Chemistry Notes For vtu students (Common to ... Subject Code 17CHE12/17CHE22 IA Marks 40 Number of Lecture Hours/Week 04 Exam Marks 60 Total Number of Lecture Hours 50 Exam Hours 03 CREDITS - 04 Course objectives: To provide students with knowledge of engineering chemistry for building technical competence in industries, research and development in the following fields □ Electrochemistry & Battery Technology. □ Corrosion & Metal ...

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